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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/758,552	01/15/2004	Jon P. Wagner	20031231-001	5985
34160	7590	12/30/2004		EXAMINER
SUD-CHEMIE INC. 1600 WEST HILL STREET LOUISVILLE, KY 40210				JOHNSON, CHRISTINA ANN
			ART UNIT	PAPER NUMBER
			1725	

DATE MAILED: 12/30/2004

Please find below and/or attached an Office communication concerning this application or proceeding.

Office Action Summary	Application No.	Applicant(s)	
	10/758,552	WAGNER ET AL.	
	Examiner	Art Unit	
	Christina Johnson	1725	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

1) Responsive to communication(s) filed on 15 January 2004.

2a) This action is FINAL. 2b) This action is non-final.

3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

4) Claim(s) 1-20 is/are pending in the application.
4a) Of the above claim(s) _____ is/are withdrawn from consideration.
5) Claim(s) _____ is/are allowed.
6) Claim(s) 1-20 is/are rejected.
7) Claim(s) _____ is/are objected to.
8) Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

9) The specification is objected to by the Examiner.

10) The drawing(s) filed on _____ is/are: a) accepted or b) objected to by the Examiner.

 Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).

 Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).

11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).

a) All b) Some * c) None of:

1. Certified copies of the priority documents have been received.
2. Certified copies of the priority documents have been received in Application No. _____.
3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

1) Notice of References Cited (PTO-892)
2) Notice of Draftsperson's Patent Drawing Review (PTO-948)
3) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08)
Paper No(s)/Mail Date ____.

4) Interview Summary (PTO-413)
Paper No(s)/Mail Date. ____.
5) Notice of Informal Patent Application (PTO-152)
6) Other: ____.

DETAILED ACTION

Claim Rejections - 35 USC § 103

1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

2. Claims 1-13 and 18-20 are rejected under 35 U.S.C. 103(a) as being unpatentable over WO 00/54879 in view of Silver.

WO 00/54879 (Note that EP 1 161 991 is an English language equivalent) discloses a catalyst composition useful in water gas shift reactions for converting carbon monoxide and water into carbon dioxide and hydrogen (page 1, lines 5-10 and 25-30). The catalyst composition comprises platinum supported on a metal oxide carrier such as zirconia (page 2, line 25 – page 3, line 5). Platinum is supported in an amount in the range of 0.1-10% by weight (page 3, lines 5-8). It is taught that rhenium is supported as another active component in an amount in the range of 0.1-10% by weight (page 3, lines 5-10). In an example, a catalyst is prepared having 3% by weight Pt and 1% by weight Re, which yields a Pt/Re ratio of 3:1 (page 14, Table 3). It is further taught that the catalyst may further contain an additional metal promoter such as cerium (page 3, lines 10-18).

The WO reference teaches that the catalyst is prepared by combining the metal oxide support with aqueous salt solutions of the active metals, followed by evaporation to dryness and calcination at a temperature in the range of 400-600 degrees C (page 3,

lines 18- page 4, line 20). The use of chloroplatinic acid hexahydrate and ammonium perrhenate are exemplified as the platinum and rhenium salt solutions (Examples 1-5 and Table 2).

The difference between the reference and the claims is that the reference does not disclose the use of a support comprising cerium oxide, specifically a support comprising cerium oxide and an additive material such as zirconium dioxide.

Silver (US 6,455,182) discloses a catalyst composition useful in the catalyzing the water-gas shift reaction, converting carbon monoxide to hydrogen (column 1, lines 35-60). The catalyst composition comprises a noble metal catalyst having a promoted support, comprising a mixed oxide of at least cerium oxide and zirconium oxide (column 2, lines 55-61). Silver teaches that while cerium promoted noble metal catalysts have been shown to be effective for promoting the water gas shift reaction, they do not sufficient activity for the shift reaction without the use of a unreasonable large reactor bed (column 2, lines 15-25). Silver teaches that, through the combination of cerium and zirconia in the support material, the activity of the composition is increased and the stability of the catalyst is improved (column 2, lines 61-66). It is taught that the zirconia is present in the range of about 50-30 mole% and the ceria is present in the range of 50-70 mole % (column 4, lines 40-45). The support may further contain an additive metal such as Pr or La, in an amount in the range of 0-10 mole % (column 3, lines 1-5 and column 4, lines 44-46).

It would have been obvious to one having ordinary skill in the art at the time the invention was made to have modified the catalyst taught by the WO reference to include

the use of a mixed cerium oxide – zirconium oxide support as taught by Silver. The WO reference teaches a catalyst composition comprising Pt and Re supported on zirconia and further suggests an additional promoter such as cerium. The teachings of Silver would motivate one of ordinary skill to substitute the zirconia carrier with the mixed cerium oxide-zirconium oxide carrier in order to realize the advantages disclosed therein, i.e. higher activity and stability. Because both catalysts can be used in the same process, one would have reasonable expectation of success from the combination.

The process limitations in claim 13 is noted. However, when the examiner has found a substantially similar product as in the applied prior art, the burden of proof is shifted to applicant to establish that their product is patentably distinct and not the examiner to show the same process of making. *In re Brown*, 173 USPQ 685 and *In re Fessmann*, 180 USPQ 324.

3. Claims 14-17 are rejected under 35 U.S.C. 103(a) as being unpatentable over WO 00/54879 in view of Silver as applied to claims 1-13 and 18-20 above, and further in view of WO 00/66486.

The teachings of WO 00/54879 as modified by Silver are applied as described above for claims 1-13 and 18-20.

The modified disclosure of WO 00/54879 further does not teach that the primary transition metal is proved as a transition metal complex having at least one ligand, where the ligand is absent of sulfur, chlorine, sodium, bromine, and iodine.

WO 00/66486 discloses a catalyst composition useful in the conversion of carbon monoxide and water into carbon dioxide and hydrogen comprising a platinum supported on zirconium oxide (Abstract). The reference teaches that while chloroplatinic acid is an inexpensive source of platinum, tetra amine platinum nitrate (TAPN) is preferred as a source of platinum because TAPN does not introduce chlorine into the catalyst system, which would interfere with the reaction. See pages 11-12.

It would have been obvious to one having ordinary skill in the art to further modify the invention of WO '879 to include the use of TAPN in light of the teachings of WO '486. One of ordinary skill would have been motivated to do so to prevent introducing a contaminant into the catalyst system, as taught by WO '486. Because both catalysts can be used in the same reaction, one would have reasonable expectation of success from the combination.

Conclusion

4. The prior art made of record and not relied upon is considered pertinent to applicant's disclosure.
5. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Christina Johnson whose telephone number is (571) 272-1176. The examiner can normally be reached on Monday-Friday, 7:30-5, with Alternate Fridays off.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Tom Dunn can be reached on (571) 272-1171. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

on phone
Christina Johnson
Patent Examiner
Art Unit 1725
12/20/04

CAJ
December 20, 2004